



# **Exploring novel catalysts for the** production of green methanol from CO,

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CO<sub>2</sub> hydrogenation to methanol



H, from electrolysis powered by PV/solar

Captured **CO**<sub>2</sub> (biomass, steel, cement industry, etc)



**S** vacancies Mo S

T > 300° C

### Molybdenum sulfide (MoS<sub>2</sub>) catalysts

**in-plane vacancies**:  $CO_2$  hydrogenation to  $CH_3OH^{[1]}$ edge vacancies: CO<sub>2</sub> hydrogenation to CH

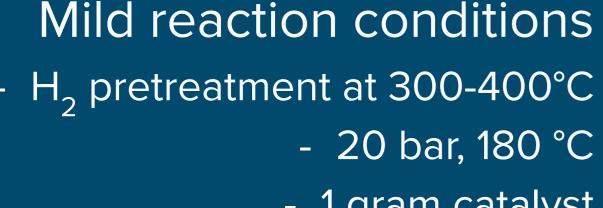
Advantages of MoS<sub>2</sub>: - Stability

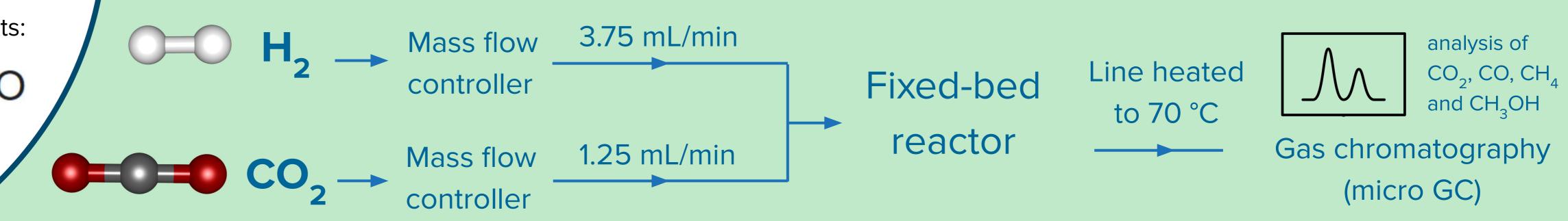
- Tolerance to sulfur
- impurities
  - Operation under
  - mild conditions<sup>[1]</sup>

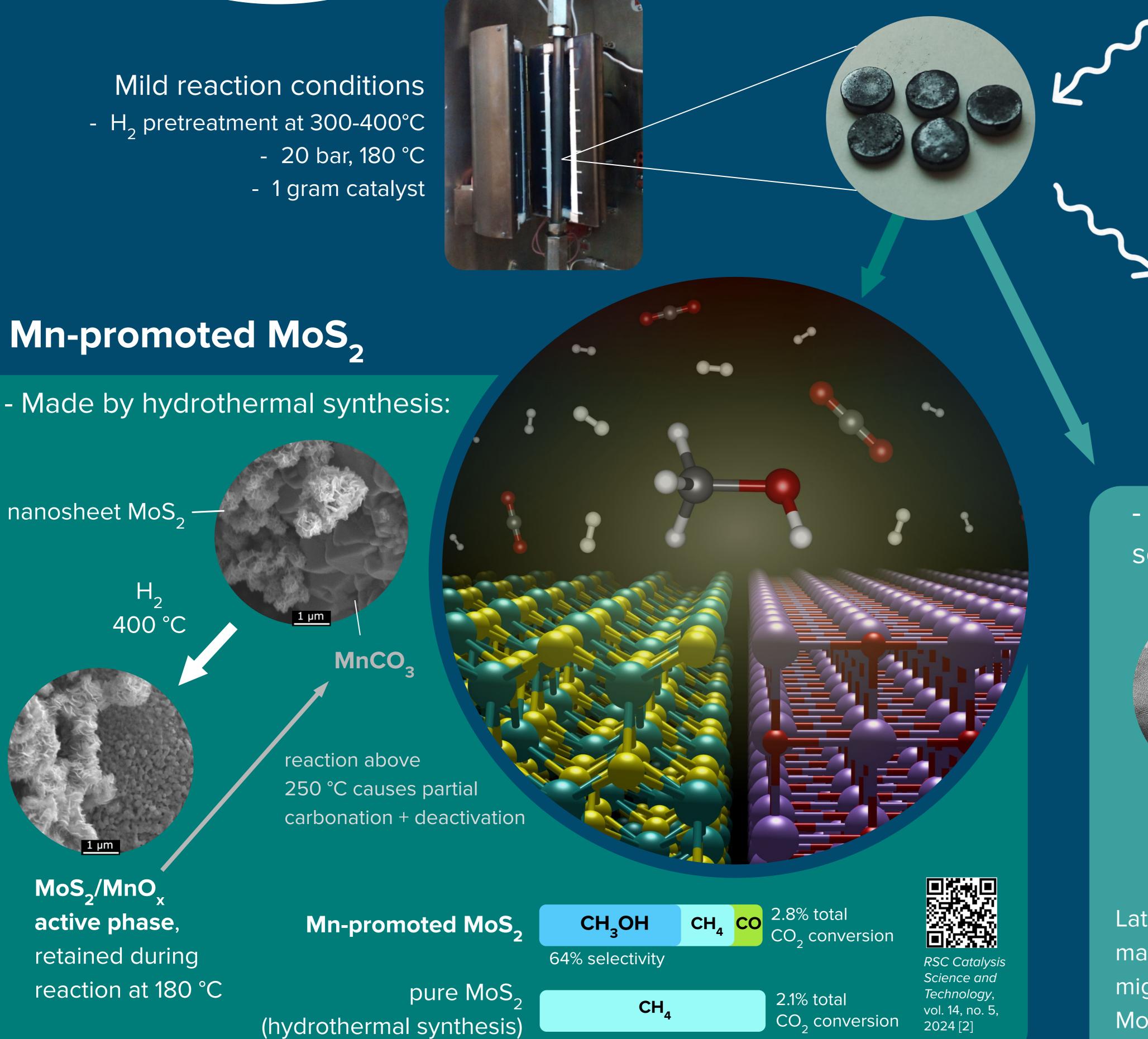
Typically ~250 °C, 50 bar, with Cu/ZnO catalysts:

 $CO_2 + 3H_2 \rightarrow CH_3OH + H_2O$ 

Sustainable alternative to the conventional fossil fuel-based methanol synthesis





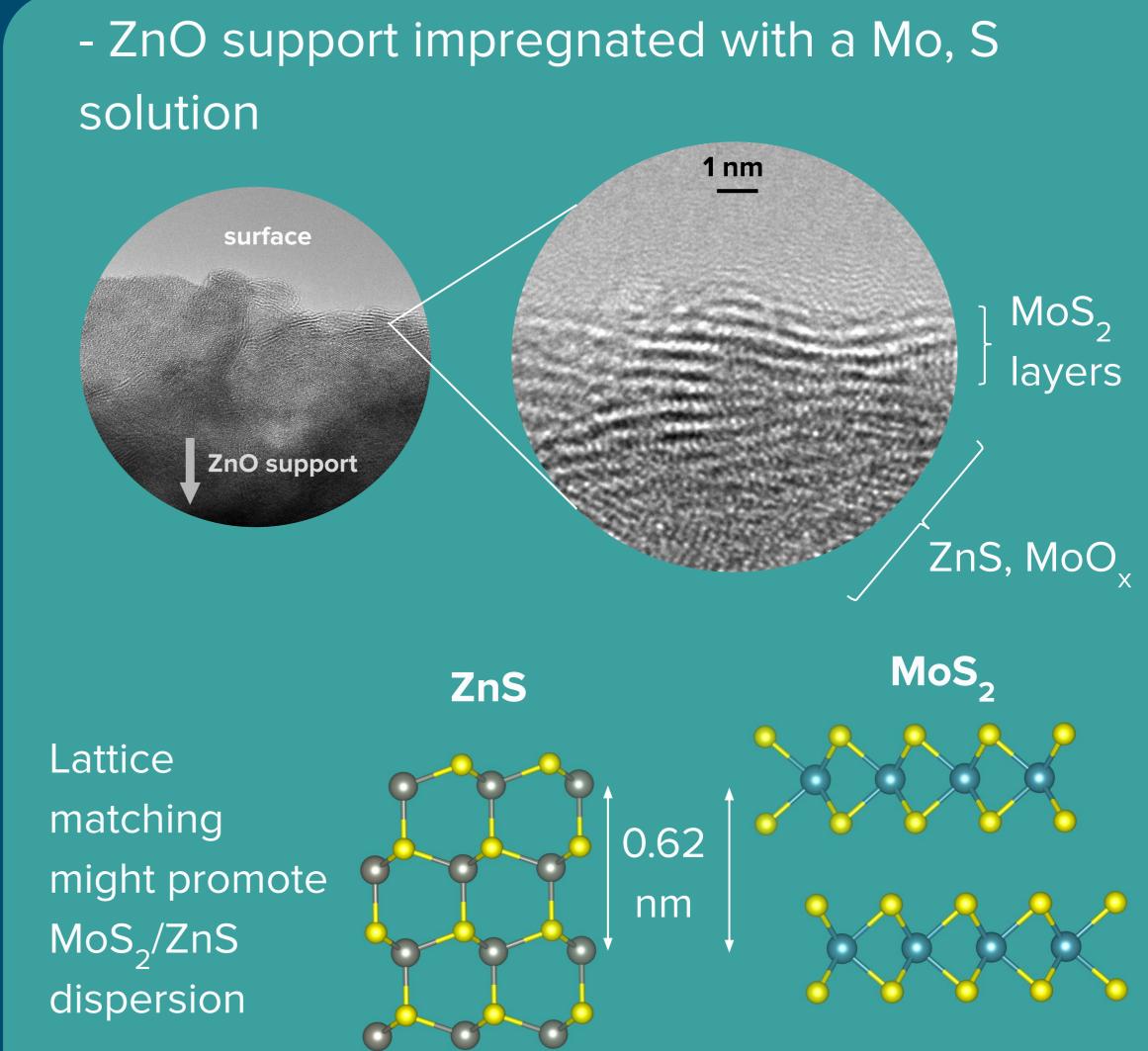


Material characterization Structure:

- X-ray diffraction (XRD)
- X-ray absorption spectroscopy (XAS)
- Transmission Electron Microscopy (TEM)
- Electron Paramagnetic Resonance (EPR) Surface:
- X-ray photoelectron spectroscopy (XPS)
- Scanning Electron Microscopy (SEM)

## ZnO-supported MoS

MoOx

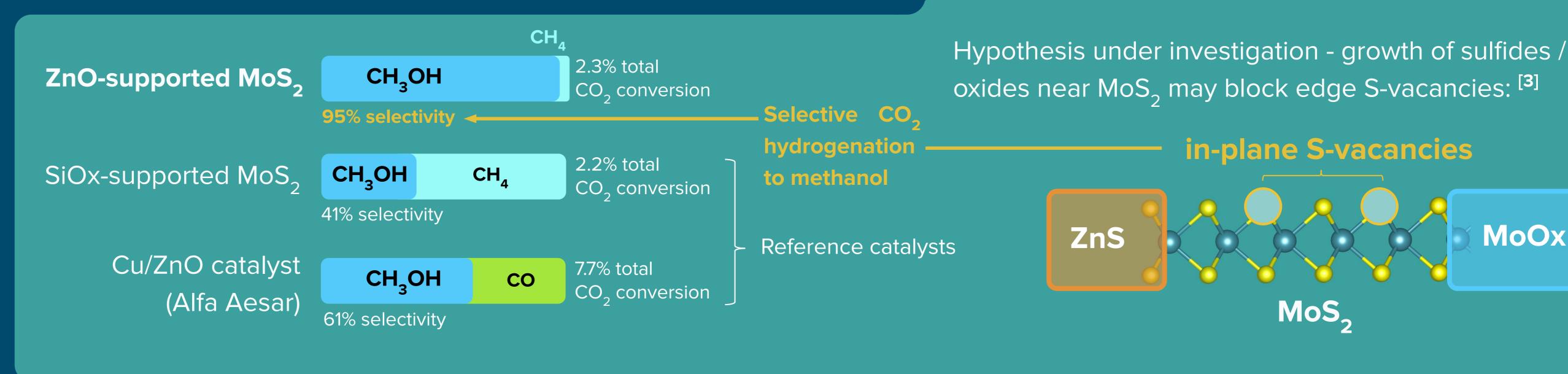




Spin-off targeting upscaling studies, experiments with industrial CO<sub>2</sub>

sources:





### **References:**

[1] Hu, J. et al. Sulfur vacancy-rich MoS<sub>2</sub> as a catalyst for the hydrogenation of CO<sub>2</sub> to methanol. Nature Catalysis (2021), 4, 242-250

[2] Alves, G.A.S. et al. Mn-promoted MoS<sub>2</sub> catalysts for CO<sub>2</sub> hydrogenation: enhanced methanol selectivity due to MoS<sub>2</sub>/MnO<sub>2</sub> interfaces. Catalysis Science & Technology (2024), 14, 5, 1138 [3] Zhou, S., Zeng, H. C. Boxlike Assemblages of Few-Layer MoS<sub>2</sub> Nanosheets with Edge Blockage for High-Efficiency Hydrogenation of CO<sub>2</sub> to Methanol. ACS Catalysis (2022), 12, 16, 9872–9886